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## **Berberine as a Photosensitizing Agent for Antitumoral Photodynamic Therapy: Insights into its Association to Low Density Lipoproteins.**

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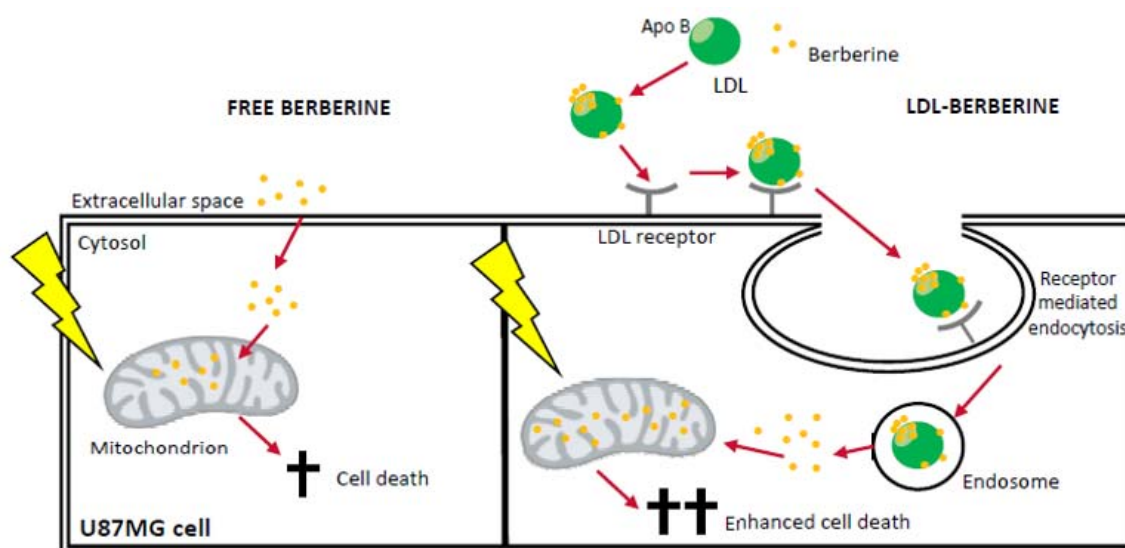
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## GRAPHICAL ABSTRACT



## ABSTRACT

Recent years have seen a growing interest in Berberine, a phytochemical with multispectrum therapeutic activities, as anti-tumoral agent for photodynamic therapy (PDT). In this context, low density lipoproteins (LDL) play a key role in the delivery of the photosensitizer in tumor cells. We correlate the physicochemical parameters of the berberine association to LDL with the influence of LDL-delivery on its accumulation in a glioma cell line and on its photo-induced activity in view of antitumor PDT. Our results evidence an important binding of 400 berberine molecules per LDL. Changes in berberine and apoprotein fluorescence suggest different fixation types, involving various LDL compartments including the vicinity of the apoprotein. The berberine association to LDL does not affect their recognition by the specific B/E receptors, of which over-expression increases the cellular uptake of LDL-preloaded berberine. Fluorescence microscopy evidences the mitochondrial labeling of the glioma model cells, with no significant modification upon LDL-delivery. Moreover, the cellular delivery of berberine by LDL increases its photocytotoxic effects on such cells. So, this research illustrates the potential of berberine as a photosensitizing agent for PDT, in particular due to their behavior towards LDL as plasma vehicles, and gives insights into its mechanisms of cell uptake.

**Keywords:** Berberine, low-density lipoprotein, antitumor PDT, cellular uptake, subcellular localization.

## 1- INTRODUCTION

Berberine (**Figure S1**) is an isoquinoline alkaloid present in species of Annonaceae and Ranunculaceae families. This natural compound is traditionally used in Chinese medicine to treat gastrointestinal diseases (Remppis et al., 2010) and has multiple pharmacological activities including anti-diabetic, anti-inflammatory, anti-malarial, anti-microbial, anti-cancer and antioxidative (Gu et al., 2010; Kuo et al., 2004; Küpeli et al., 2002; Sarna et al., 2010; Vuddanda et al., 2010; Wang et al., 2009; Wu et al., 2010; Yin et al., 2008). In addition,

berberine has been confirmed to penetrate the blood-brain barrier and exhibits beneficial effects in the central nervous system (Kulkarni and Dhir, 2010; Lin et al., 2013).

In the last ten years, berberine has been more extensively studied as photosensitizing agent for photodynamic therapy (PDT) (L. Cheng et al., 2009; L.-L. Cheng et al., 2009; Inbaraj et al., 2001; Jantová et al., 2006). PDT is an attractive alternative antitumor approach, due to both its specificity and its ability to be combined with photodynamic diagnosis imaging (PDD), which presents a great interest largely highlighted in the recent years (Jayaram et al., 2016; Silva et al., 2013). The concomitant actions of a photo-activable substance, so-called photosensitizer (PS), light and oxygen is the basic principle of this approach (Castano et al., 2004). The search for efficient sensitizers with high specificity has provided potential impetus because of the wide range of applications of the antitumor PDT (Acherar et al., 2015; Pariser et al., 2003; Peng et al., 2009), that is promising for new applications such as in urology (Azzouzi et al., 2013), pneumonology (Du et al., 2010) or neurosurgery, where encouraging results have been reported concerning the use of interstitial photodynamic therapy as treatment of front-line nonresectable or recurrent glioblastoma (Beck et al., 2007).

Photodynamic therapy involves in general a systemic administration of the photosensitizer to the patient, followed by the light irradiation of the targeted zone. As the photo-induced cytotoxic species have very short lifetimes and subsequently limited diffusions (Kerdous et al., 2011; Kuimova et al., 2009; Moan and Berg, 1991), primary damages affect the subcellular structures labeled by the photosensitizer. Its intracellular distribution is then a major determinant of the PDT efficacy, although the overall response to these primary damages is complex (Piette et al., 2003). As a matter of fact, note that the subcellular localization of photosensitizers is well correlated with their global photocytotoxicity (Woodburn et al., 1992) and the mechanism of cell death (Hsieh et al., 2003; Kessel et al., 1997). The relative photo-efficiency and the subcellular localization of photosensitizers depend on their structure, in particular on their hydrophobicity and charges (Bonneau et al., 2007; Boyle and Dolphin, 1996; Hsieh et al., 2010).

Among the processes that could account for the retention of photosensitizers by tumor tissues, an important one is related to their hydrophobicity and is specific to these proliferating targets. In tumor cells, indeed, the increase of the cholesterol catabolism results in an over-expression of low-density lipoproteins' receptors (B/E receptors). Moreover, as lipophilic drugs, the photosensitizers present a high affinity for low-density lipoproteins (LDL) (Bonneau and Vever-Bizet, 2008; Huntosova et al., 2010). Hence, as previously shown for different lipophilic and amphiphilic photosensitizers, LDL can act as targeted nano-carriers enhancing the pharmacological efficacy of these agents (Bonneau et al., 2004; Huntosova et al., 2010; Mojzisoava et al., 2007). Besides, the LDL ability to be specifically transcytosed across the blood-brain barrier (Candela et al., 2008) must be taken into consideration regarding the encouraging promises of PDT in neurosurgery.

In this context, the first aim of this study is to describe the physicochemical properties of the berberine interaction with LDL, using spectroscopic methods. Different classes of binding have been identified and characterized by monitoring both the fluorescence of LDL and that of berberine. We have then evaluated the effect of this interaction on the drug behavior toward human primary glioblastoma cell line. The cellular uptake and subcellular localization in cell models with standard (normal) and enhanced number of LDL receptors were analyzed

and, finally, the photosensitizing potential of both free and LDL-associated berberine determined through lipid peroxidation and cell viability assays.

## 2- MATERIAL AND METHODS

### 2.1- Chemicals.

Berberine (BBR, 99% purity), dimethyl sulfoxide (DMSO, 99.9%) and Human Serum Albumin (HSA), essentially fatty acid free, were purchased from Sigma-Aldrich (Saint Louis, Missouri, USA). The chlorin e6 (Ce6) was purchased from Frontier Scientific (Logan UT, USA). Ethanol (99.8%) was purchased from Merck (Darmstadt, Germany). Phosphate buffer Saline (PBS) and Hank's balanced salt solution (HBSS), both at pH 7.2 and all the other cell culture reagents were purchased from GIBCO-Invitrogen (Cergy Pontoise, France), except the Ultrosor G obtained from BioSeptra (SA-Part of Pall Corporation, France). Human low density lipoproteins (LDL, purity >95%) were purchased from Calbiochem (Nottingham, England) and the fluorescent probes Mitotracker® Red and LysoTracker® Red from Molecular Probes (Eugene, Oregon, USA). The reagents for all biochemical analyses were purchased from Sigma-Aldrich (Saint Louis, Missouri, USA). The storage proceedings of all these chemicals, reagents and substances were made according to manufacturer's information. The photosensitizer solutions were handled in dark to avoid any photobleaching.

### 2.2- Spectroscopic studies.

The absorption spectra were measured with an UVIKON 923 double beam UV/VIS spectrophotometer (BioTEK Kontron Instruments). Fluorescence emission spectra were recorded using an AMINCO-Bowman 2 spectrofluorimeter (Edison, NJ) equipped with a xenon arc lamp. Solutions were in PBS pH 7.2, otherwise further information is given. Recording was generally started 2 min after solution's preparation. Data thus obtained were analyzed as briefly described in the subsequent paragraph.

### 2.3- Analysis of the interaction of LDL with berberine

Partition experiments of berberine between LDL and aqueous solution were performed at equilibrium with various LDL concentrations (0 to  $10^{-8}$  M). The global binding constant,  $K_{LDL}$ , was derived from the changes in the berberine fluorescence at 544 nm (corresponding to the maximum of the fluorescence emission of berberine within LDL). We used the previously derived relationship (Kuzelova and Brault, 1994):

$$F - F_0 = \frac{(F_\infty - F_0) \times K_{LDL} \times [LDL]}{1 + K_{LDL} \times [LDL]} \quad (1)$$

where  $F_0$ ,  $F_\infty$  and  $F$  are the fluorescence intensities of berberine corresponding to zero, total and intermediate association of the berberine to LDL, respectively.

According to our experimental results, the berberine association to LDL was described as involving two class of fixations. The first class, denoted *class P*, corresponds to the binding of berberine molecules evidenced by the quenching of the fluorescence of the apoprotein B100 of LDL. From this binding-induced changes in LDL-fluorescence properties, we derived the association parameters with the previously described method according with (Halfman and Nishida, 1972) and (Bonneau et al., 2004, 2002) as follows.

By definition, at a given LDL concentration:

$$P_F = LDL (P_T/LDL - \nu) \quad (2)$$

where  $P_F$  and  $P_T$  are respectively the concentrations of free berberine in the aqueous solution and the total berberine concentration,  $\nu$  the number of berberine molecules bound per lipoprotein, and LDL the concentration of the latter. The Nishida method is based on the existence of pairs of  $P_T$  and LDL concentrations yielding the same value of  $\nu$ . Then, for two pairs noted a and b it follows from equation (2):

$$\nu = [P_{Ta}/LDL_a - (LDL_b/LDL_a) * (P_{Tb}/LDL_b)] / (1 - LDL_b/LDL_a) \quad (3)$$

$$P_F = [LDL_a LDL_b / (LDL_a - LDL_b)] (P_{Tb}/LDL_b - P_{Ta}/LDL_b) \quad (4)$$

Assuming that the interaction of the same number of molecules per LDL (the same  $\nu$  value) involves the same changes in LDL fluorescence properties, these two pairs correspond to the same  $\Delta(F/F_0)$  where  $F$  and  $F_0$  are the fluorescence intensities of LDL in presence and in absence of berberine. The quenching efficiency was thus measured, for two concentrations of LDL (2 and  $6 \times 10^{-8}$  M), as a function of the total berberine concentration. Pairs of  $P_T$  and LDL concentrations yielding the same value of  $F/F_0$  were selected, and the values of  $\nu$  and  $P_F$  were derived according to equation (3) and (4). The results were plotted according to the Scatchard method.

The binding of additional berberine molecules to LDL, which do not change the intrinsic fluorescence of LDL, correspond to the second class of fixation and are denoted *class L*.

#### 2.4- Cell culture and experimental series preparation.

Cells from the human primary glioblastoma cell line U87-MG were obtained from the American Type Culture Collection (Manassas, VA, USA) and routinely grown as monolayer at 37°C in Dulbecco's modified Eagle's medium (DMEM) supplemented with 10% fetal bovine serum (FBS) and 100 U/ml of penicillin and 100 µg/ml of streptomycin, under atmosphere controlled at 100% of relative humidity and 5% CO<sub>2</sub>. The cells were passed every 4 days and used between the 10<sup>th</sup> and the 19<sup>th</sup> passages. Experimental series are defined as follows: cells are incubated with berberine (LDL- series) or with the LDL-Berberine complex (LDL+ series) in HBSS at 37°C. For the experiments on cells with over-expression of LDL receptors (R+ series), FBS was replaced within the growing medium by 2% Ultrosor G one week before experiments (Mojzisova et al., 2007). For all the series (R-/LDL-, R-/LDL+, R+/LDL- and R+/LDL+), the berberine is used at  $1 \times 10^{-6}$  M and LDL at  $1 \times 10^{-8}$  M LDL (BBR/LDL=100). A fifth series, used as control, involved experiments carried out by using the same protocol but without any photosensitizer.

#### 2.5- Microscopy.

Fluorescence images were obtained using an inverted microscope (Nikon Eclipse TE300) equipped with a high aperture phase oil objective (CFI Plan apochromat DM 60x N.A. 1.4, Nikon, France). The excitation light was provided by a 130W mercury lamp illuminator (Intensilight, Nikon). Illumination wavelengths were isolated by band-pass filters ( $355 \pm 25$  nm for berberine,  $570 \pm 25$  nm for the used organelles probes) and intensity was optimized by neutral density filters (ND4 and ND8). The fluorescence emission was collected through band-pass filters ( $535 \pm 25$  nm for berberine and  $610 \pm 10$  nm for the organelles probes). The images were acquired with a camera Neo sCMOS (Andor Technology). The acquisition, processing and image analysis were performed with NIS-Element provided by Nikon and Image J (open source software, <http://rsb.info.nih.gov/ij/index.html>).

### 2.6- Berberine uptake by U87- MG cells.

After incubation for 15, 30, 60, 120 and 180 minutes, to quantify the uptake by extraction, the cells seeded in Petri dishes were used at 70% confluence. After incubation and washing, they were scrapped in 900  $\mu$ l H<sub>2</sub>O and 100  $\mu$ l of 3% Triton X100 solution. Then, 100  $\mu$ l of the disrupted cell solution were saved for the protein determination by Lowry's method. The 900  $\mu$ l of remaining solution was used for the fluorimetric measurement of the berberine concentrations. Data, expressed as  $\mu$ mole of berberine per gram of protein, are means ( $\pm$ SD) of triplicates. Alternatively, after one hour of incubation, cells were observed under fluorescence microscope.

### 2.7- Subcellular localization of berberine within U87-MG cells.

After one hour of incubation, the cells were washed twice with HBSS and then incubated with Mitotracker<sup>®</sup> Red or LysoTracker<sup>®</sup> Red (200 nM) for 30 minutes. The Pearson Coefficient (CP) was determined using Image J.

### 2.8- Irradiation set-up.

Irradiation was carried out using a 300 W xenon lamp. A 410 $\pm$ 10 nm band-pass filter (Andover Corporation) was used to select the irradiation band. As positive control, we used samples incubated with chlorin e6, a well known photosensitizer (Mojzisova et al., 2009, 2007), irradiated through a K65 Balzer band-pass filter (660 $\pm$ 15 nm). The light beam had a diameter of about 50 mm, allowing the irradiation of a 30 mm diameter plate at a time. The irradiance uniformity was controlled by scanning the zone with a power meter. Variation of intensity observed was less than 5%. The liquid height in the wells was only 0.5 cm to prevent light attenuation. For the light-dose dependent studies, the samples were irradiated at light fluencies of 10, 25, 50 and 100 J/cm<sup>2</sup> (65 $\pm$ 0.5 mW/cm<sup>2</sup> for 2.5, 6.5, 13 and 23 min, respectively).

### 2.9- Evaluation of the photo-induced effects on cells.

Lipid peroxidation within irradiated cells was evaluated by TBARS assay, both on cellular medium culture and cell extract. The reagent solution was added to the same volume of the sample of interest. The mixture was heated in a water bath at 80°C for 15 min, butanol was then added, and the organic phase was analyzed by spectrofluorometry. Spectra were recorded between 525 and 800 nm with an excitation at 415 nm. Data are means ( $\pm$ SD) of triplicates in three different assays.

The viability of irradiated cells was assessed by colorimetric MTT assays. Cells were plated into 96-well plates at a density of 5 $\times$ 10<sup>4</sup> cells/well one day before experiments. After treatment and washing twice, MTT solution (5 mg/ml) and medium were introduced. After incubation for 4 h, the formed formazan crystals were dissolved in dimethyl sulfoxide and the absorbance intensity measured by a microplate reader (Synergy 2, BioTek Instruments, USA). Each sample was evaluated in triplicate and the experiment was repeated three times. Percentage of cell proliferation was calculated as the ratio between the absorbance at 570 nm of the sample and that of the control. Data are means ( $\pm$ SD) of triplicates in three different essays.

### 2.10- Statistical analysis.

All data are means ( $\pm$ SD) of at least three independent replicates and were processed and represented using GraphPad Prism software (GraphPad Software, version 5.00, San Diego,

CA). Comparisons among the different series of each experiment were performed by analysis of variance and Tukey-Kramer range test. Correlations were analyzed with Pearson correlation analysis. P values <0.05 were considered significant. Letters indicate if the compared samples are statistically different (represented by different letters) or not (represented by equal letters) within the evaluated group.

### 3- RESULTS AND DISCUSSION

#### 3.1- Spectroscopic properties of berberine.

The berberine absorption and fluorescence spectra were recorded at different concentrations in solvents with different physicochemical properties (**Figure S1**). The absorption spectra exhibit only small changes depending on the solvent. The absorption bands are slightly blue-shifted ( $17\pm 2$  nm) with increasing solvent polarity. The fluorescence is more sensitive to the environment. In organic solvents, the maximum intensity is halved and the corresponding wavelength red-shifted with polarity decrease. In PBS, berberine exhibits a drastic fluorescence decrease (max 544 nm). Our observations (**Table S1**) are in accordance with previous ones (L. Cheng et al., 2009; L.-L. Cheng et al., 2009; Jantová et al., 2006).

#### 3.2- The berberine interaction with LDL.

The interaction between berberine and LDL was primarily evidenced by the changes of berberine fluorescence in presence of LDL (**Figure 1**). The intensity is strongly enhanced and shifted from 544 to 510 nm, with a behavior similar to that observed in organic solvents. Moreover, the UV-Vis absorption spectrum of LDL in complexed form (subtraction of the berberine absorption to that of the mixture) cannot be superposed with that of naive ones, suggesting an association of both species. The alkaloid binding has then been studied by assessing induced changes in both the berberine fluorescence and the LDL one.

First, fluorescence measurements were performed by adding increasing amounts of LDL to berberine solution ( $10^{-6}$  M) and the berberine fluorescence was simultaneously followed (**Figure 2**). The plot of the fluorescence intensity presents a plateau, of which the intercept with the initial slope corresponds to an LDL concentration of  $2.5\times 10^{-9}$  M. The corresponding ratio of concentrations between bound berberine and LDL indicates that each lipoprotein could bind about 400 berberine molecules at best. Analysis according to equilibrium equation (1) gives a global association constant ( $K_{LDL}$ ) of  $(1.76\pm 0.2)\times 10^8$  M<sup>-1</sup>. Duplication of experiments with other berberine and LDL concentrations led to consistent values.

In a second step, the binding characterization was apprehended by following the changes of the LDL fluorescence induced upon berberine addition. Indeed, under excitation at 280 nm, the LDL display an intrinsic fluorescence emission band near 330 nm, due to the tryptophan residues of the apoprotein B100. The significant overlap of this band with the berberine absorption spectrum allows a Förster Resonance Energy Transfer (FRET) to the alkaloid. The Förster distance has been estimated to be 3,1 nm (Hu et al., 2010) for a related berberine/tryptophan couple. The binding of berberine molecules at the direct vicinity of the apoprotein (to the apoprotein or at the frontier between protein and lipids) may then be assessed by FRET. Such associations will be denoted *class P* fixation.

The FRET efficiency is experimentally evidenced by the quenching of the apoprotein fluorescence by the binding of berberine, in a concentration-dependent manner (**Figure 3A**) with a concomitant fluorescence increase around 500 nm. The significant fluorescence



quenching observed even for relatively low amounts of alkaloid indicates the presence of high-affinity *class P* sites. The blue-shift of the LDL fluorescence suggests an increased hydrophobicity of the environment of the non-quenched tryptophan residues. For low degree of berberine binding, a linear stoichiometric relationship between the fluorescence intensity and the degree of binding was observed. The Stern-Volmer constant has been calculated to be  $(1.11 \pm 0.05) \times 10^5 \text{ M}^{-1}$  (**Figure 3B**). For higher degree of berberine binding, the relationship depends upon the protein concentration (**Figure 3C**). Indeed, for each site, the FRET efficiency depends on the distance and relative orientation of the donor-acceptor couple, and its dependency on the number of berberines bound per LDL is not linear. Hence, the binding related to this quenching has been characterized using the method developed by Halfman and Nishida (Halfman and Nishida, 1972). This approach is based on the existence of pairs of berberine and LDL concentrations yielding to the same number of berberine molecules bound per lipoprotein and a unique quenching efficiency, as described elsewhere (Bonneau et al., 2002). According to calculations briefly described in the section 2.3, two sets of sites appear to adequately describe this binding process (**Figure 3D**). The first set, denoted *class P type 1*, involves 180 sites of which intrinsic affinity,  $K_{P1i}$ , is  $8.6 \times 10^5 \text{ M}^{-1}$ . The second set of *class P type 2* fixations is identified as 40 sites of intrinsic affinity of  $2.5 \times 10^5 \text{ M}^{-1}$ . The total association constant for these two types of *class P* sites, i.e. identified by the induced changes in LDL fluorescence and thus involving localizations at the vicinity of the apoprotein, is then  $1.65 \times 10^8 \text{ M}^{-1}$ . The precise localization of these two types of site will be discussed with respect to the LDL recognition by their cellular receptors. Note that the apoprotein is organized around the LDL-particle and lies in a hydrophobic environment offering possible binding areas for berberine molecules.

The first described study, in which the berberine fluorescence was followed, showed that the binding capacity exceeds that of the 220 *class P* sites identified at the vicinity of the apoprotein. Other LDL compartments including the outer phospholipid layer and the inner core should thus be involved in the binding for around 180 sites - denoted *class L* fixation - and a lower association constant,  $K_L$ , of about  $1.1 \times 10^7 \text{ M}^{-1}$ . Nevertheless, when LDL concentrations are large enough to bind all the berberine molecules, the fluorescence spectra of the latter is typical of hydrophobic environments, regardless of the  $[\text{berberine}]/[\text{LDL}]$  ratio. This indicates that the characteristics of the berberine bound to the lipoproteins in the different locations are almost similar. All the constants are summarized in the **Table 1**.

### 3.3- Berberine uptake by human primary glioblastoma cell line.

After incubation of cells with berberine solution, the fluorescence of the alkaloid is found to present a coarse granular filamentous staining (**Figure 4**), suggesting mitochondria involvement. The berberine distribution and uptake seem not to be modified upon vectorization by the lipoproteins nor by the over-expression of the LDL-receptors. To decipher the behavior of the berberine toward glioma model cells, four series of experiments have been performed, with free or LDL-preloaded berberine (LDL- and LDL+ series), on cells over-expressing or not the specific LDL receptors (R+ and R- series, see the methods section).

The uptake has been quantified by extraction. For all experiment series, the maximum of uptake appears after 60-120 minutes of incubation. All the kinetics are very similar, in good agreement with our preliminary microscopy observations, except for R+/LDL+ series. For the three others, an important decrease of the accumulated berberine takes place after 60

minutes. For the R+/LDL+ series, the maximum accumulation appears after 120 min and is still almost stable after 180 min. However, similar trends of reduction in the intracellular berberine amount can always be detected. For all series, this may indicate a possible early process of alkaloid metabolism which could, in the PDT context, be beneficial to avoid prolonged periods of light sensitivity.

Moreover, for the R+/LDL+ series, the intracellular amount of berberine is higher than those observed for all other series (**Figure 4**). Thereby, the stimulation of LDL receptors expression actually ensures an increase of the cellular uptake of the LDL-preloaded berberine up to 29% after 120 min of incubation. This indicates that the recognition of the LDL by cells is unaltered upon the binding of berberine and support the hypothesis that berberine rather interacts at the boundary between the apoprotein and the lipid phase of the LDL than at the apoprotein surface. To support this idea, comparison can be made on the basis of the number of phospholipids per LDL. The affinity constant per lipid is calculated to be  $2.2 \times 10^5 \text{ M}^{-1}$ , in good agreement with that of other photosensitizers (Bonneau et al., 2002).

### **3.4- Subcellular localization of berberine within U87-MG cells.**

Analysis of the colocalization studies of the intracellular berberine with Mitotracker Red<sup>®</sup> clearly demonstrates the mitochondrial localization of the berberine in the four series of experiments (**Figure 5**), in good agreement with previous studies (Inbaraj et al., 2001; Lakowicz, 2006; Mikes and Dadák, 1983). The average Pearson coefficient value obtained was respectively 0.94 and 0.95 for R-/LDL+ and R+/LDL+ series. No specific labeling was observed in other cytosolic organelles or plasma membrane. For images obtained with LysoTracker Red<sup>®</sup> (data not shown), the Pearson coefficient values obtained were much lower in all cases (0.2-0.3). No significant amounts of berberine have been evidenced in lysosomes (even for the LDL+ series at the shorter incubation times) although the LDL clearly play a role in the uptake of berberine, especially for the R+/LDL+ series. Taken together, our results show that LDL play a role in the cellular incorporation of the berberine, but have only a marginal effect on its intracellular behavior. This indicates a fast redistribution of the berberine in favor of mitochondria due to its release from LDL or its escape from early endocytosis compartments, during the first 15 minutes. Such behavior, leading to a specific mitochondrial accumulation, is almost classical for hydrophobic compound bearing cationic charges (Kessel et al., 2011, 2003; Ma et al., 2013; Rodriguez et al., 2009).

### **3.5- Evaluation of the photo-induced effects on U87-MG cells.**

The level of lipid oxidation in all the experiment series has been assessed by the dosage of malondialdehyde (MDA), a secondary product of lipid peroxidation reactions (**Figures 6A-D**). The photocytotoxic activity of the berberine is clearly evidenced by the light-dependent MDA production in all series in comparison with the control cells. This photooxidation increases with the light-dose. For R- series, no significant differences are induced by the LDL-preloading of the berberine. For R+ series, in contrast, the MDA production is significantly increased by the LDL-delivery for high light doses (50 and 100J/cm<sup>2</sup>). This observation is in good accordance with the fact that LDL play a role in the uptake of berberine, even if the trend to increased uptake is certainly not sufficient to explain such an increase of photocytotoxicity. However, due to the potentially different pathways, free and LDL-preloaded berberine molecules could not have similar molecular targets<sup>28</sup> and other components such as proteins can also be oxidized.

The photo-induced potential of free and LDL-associated berberine as antiproliferative agents has been tested. In all the experiment series, the cellular viability was assessed using MTT assay (**Figure 6E-F**). Both forms of berberine significantly decreased the glioma cells viability in comparison to control treatments. In both culture conditions, cell death is light-dose-dependent. However, as expected, cell viability was significantly more affected for R+/LDL+ cells. Altogether, our “upon irradiation” results show that the decrease of the viability of berberine-photosensitized cells is directly correlated to the lipid peroxidation level that appears to be well correlated to the berberine intracellular accumulation level.

#### **4- Conclusions**

Berberine presents a wide range of pharmacological applications and has driven an increasing interest, in particular for application in anticancer therapy. In this context, the present study gives insights into the role of LDL particles as natural nano-carriers of this compound. First, the binding of the berberine is important, corresponding to the fixation of 400 berberine molecules per LDL with high affinity. At physiologically relevant LDL concentration ( $7 \times 10^{-6}$  M), the ratio of associated berberine is 99.9%, what underlie the importance of their role evidenced in this work. Moreover, due to the capability of the LDL particles to cross the blood-brain barrier (Candela et al., 2008), this very high affinity suggests that the association of berberine to these lipoproteins can be involved in the berberine ability to penetrate this barrier and exhibit therapeutic potential for central nervous system disorders (Kulkarni and Dhir, 2010; Lin et al., 2013). Second, this association involves the apoprotein B100 or, most likely, its vicinity. However, the recognition of LDL by their cellular receptor is shown to be unaltered upon the binding of berberine. This suggests that the interaction is not likely to involve the outer domain of the apoprotein, as previously shown for other photosensitizers (Bonneau et al., 2007, 2002). As a matter of fact, LDL are able to increase the uptake of berberine by cells and LDL delivery of berberine allows a more specific targeting of cells characterized by a high level of B/E receptors such as in tumor cells. Moreover, delivered by LDL or not, the subcellular localization of berberine remains unchanged, in mitochondria. This may indicate a release of the drug from LDL after cellular binding or a subcellular redistribution after the cellular internalization. This point has to be taken into consideration with regards to the possible degradation of certain photosensitizers in lysosomes after endocytosis and their subsequent lack of activity. Moreover, the mitochondria are important targets in photodynamic therapy and many other hydrophobic cationic compounds have been designed to present such a behavior leading to this subcellular localization (Kessel et al., 2011, 2003; Rodriguez et al., 2009). This point is well correlated with our results showing an increased photo-oxidation and a photo-induced decreased of glioma cell viability mediated by the LDL-preloaded berberine when targeted cells over-express receptor B/E.

In summary, our results show the potential of berberine as a photosensitizing agent for PDT applications, in particular due to their behavior towards LDL as plasma vehicles. In this regard other studies are being conducted in order to precise the mechanism of cell death photo-activated by berberine once delivered by LDL.

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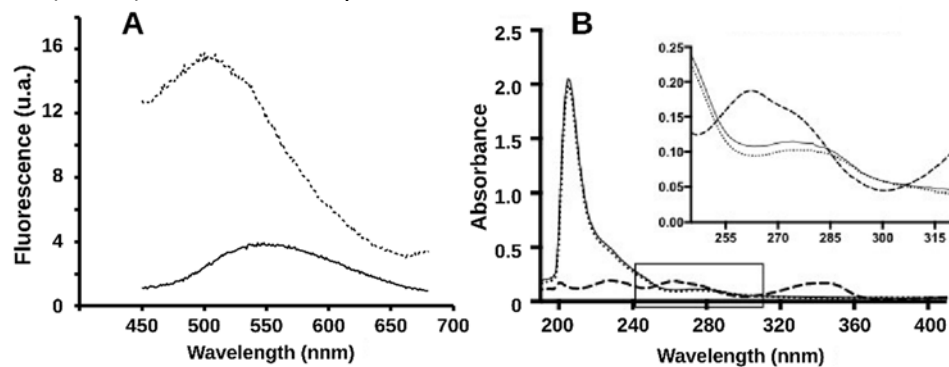
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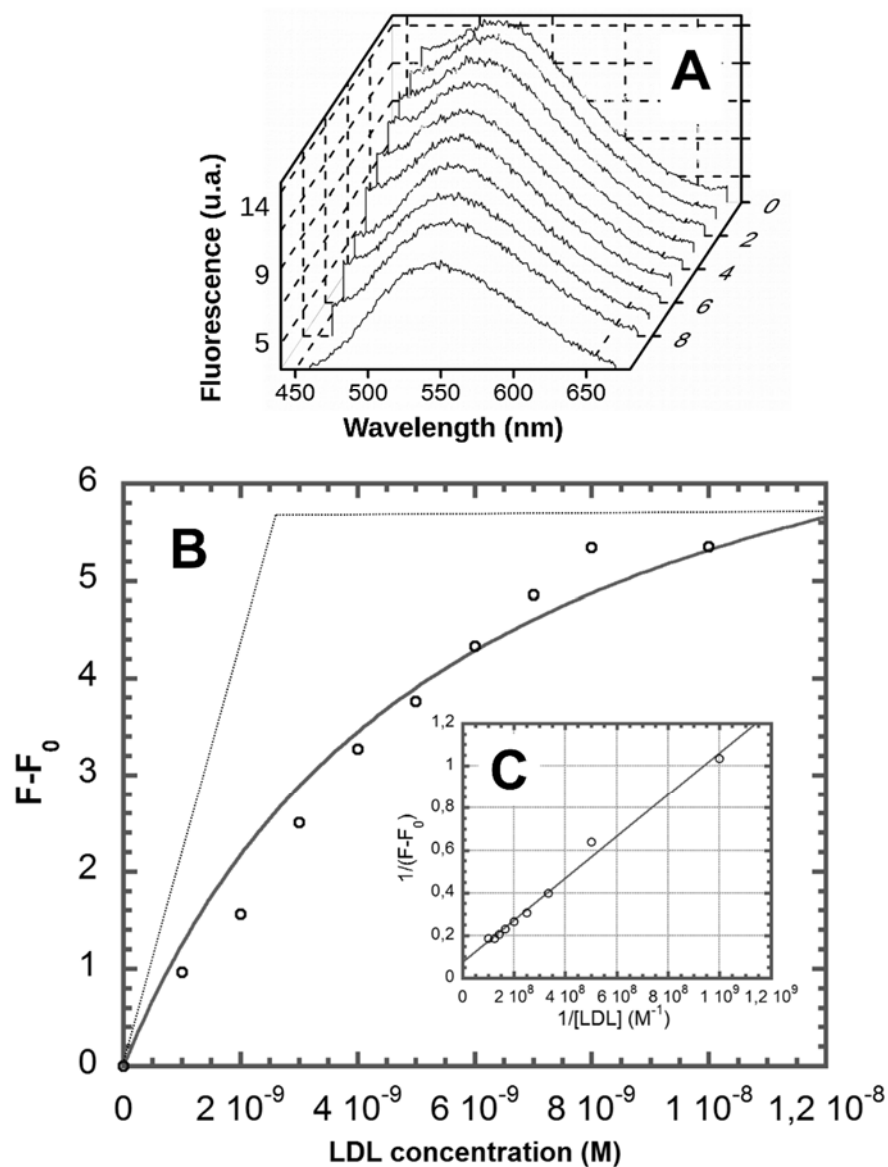
## Figure legends:

**Figure 1: Spectra of berberine with LDL.** (A) Berberine fluorescence in PBS (solid line) and in presence of LDL (dashed line). Excitation 350 nm. (B) Absorption spectrum of naive LDL (short dash), of berberine (long dash) and difference absorption spectrum between the LDL/Berberine complex and that of berberine (solid line). [LDL]=  $6 \times 10^{-8}$  M and [BBR] =  $3 \times 10^{-6}$  M. (*Insert*) Focus on the spectra from 240 to 360 nm.

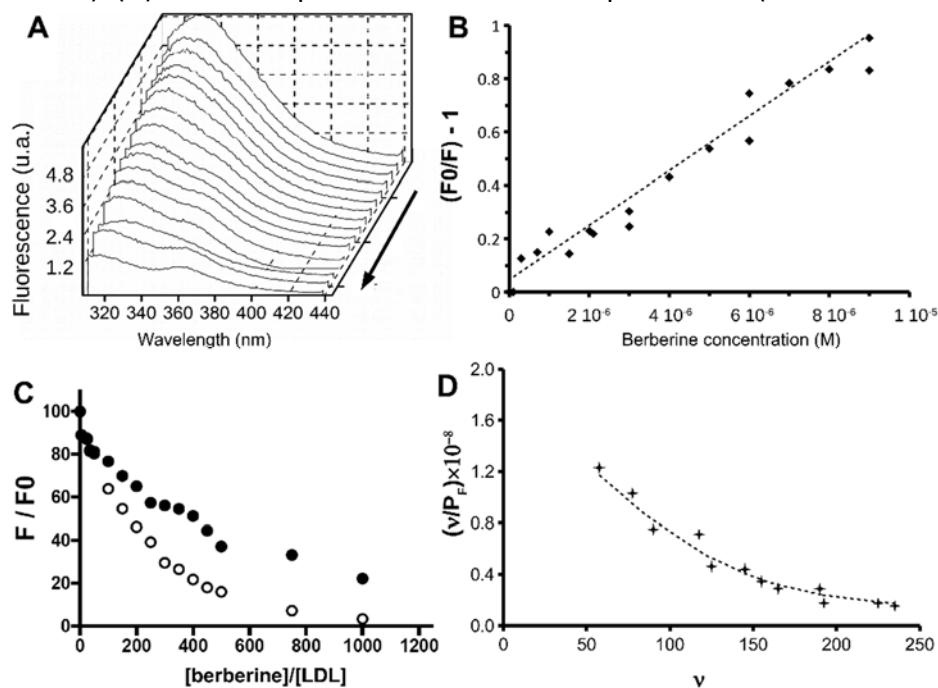




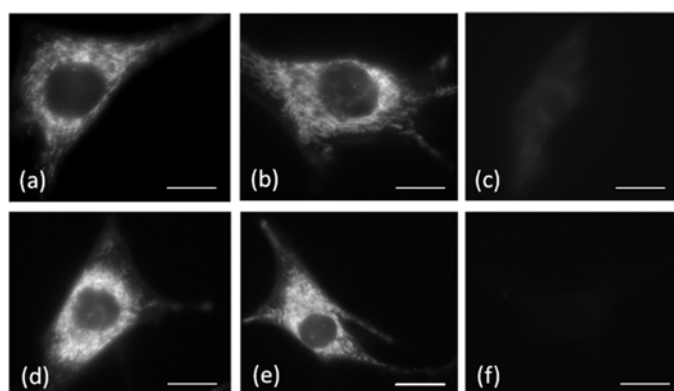
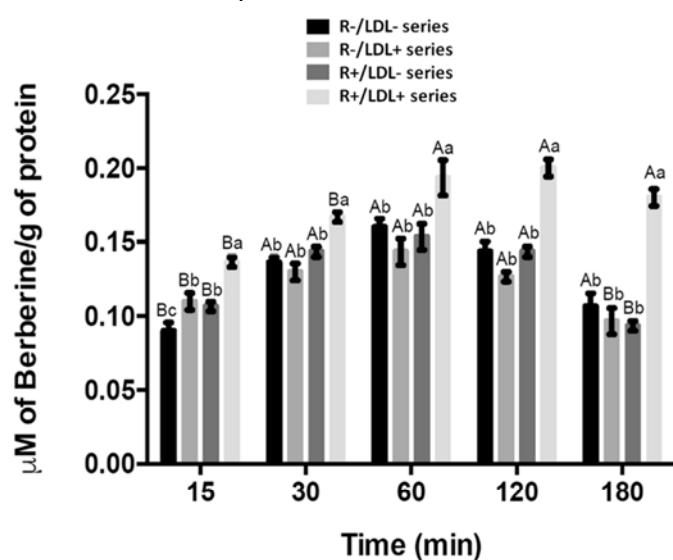
**Figure 2: Quantification of the berberine binding to LDL followed by changes of berberine fluorescence** (excitation 350 nm). (A) Fluorescence spectra of berberine ( $10^{-6}$  M) in the presence of increasing amounts of LDL (0 to  $10^{-8}$  M, spectra 10 to 0). (B) Fluorescence intensity of berberine vs. LDL concentration.  $F$  and  $F_0$  are the fluorescence intensities with and without LDL, respectively. (C) Inverse plot. The dark lines correspond to the best fit of experimental datas.  $K_{LDL} = 1.76 \times 10^8 \text{ M}^{-1}$ .



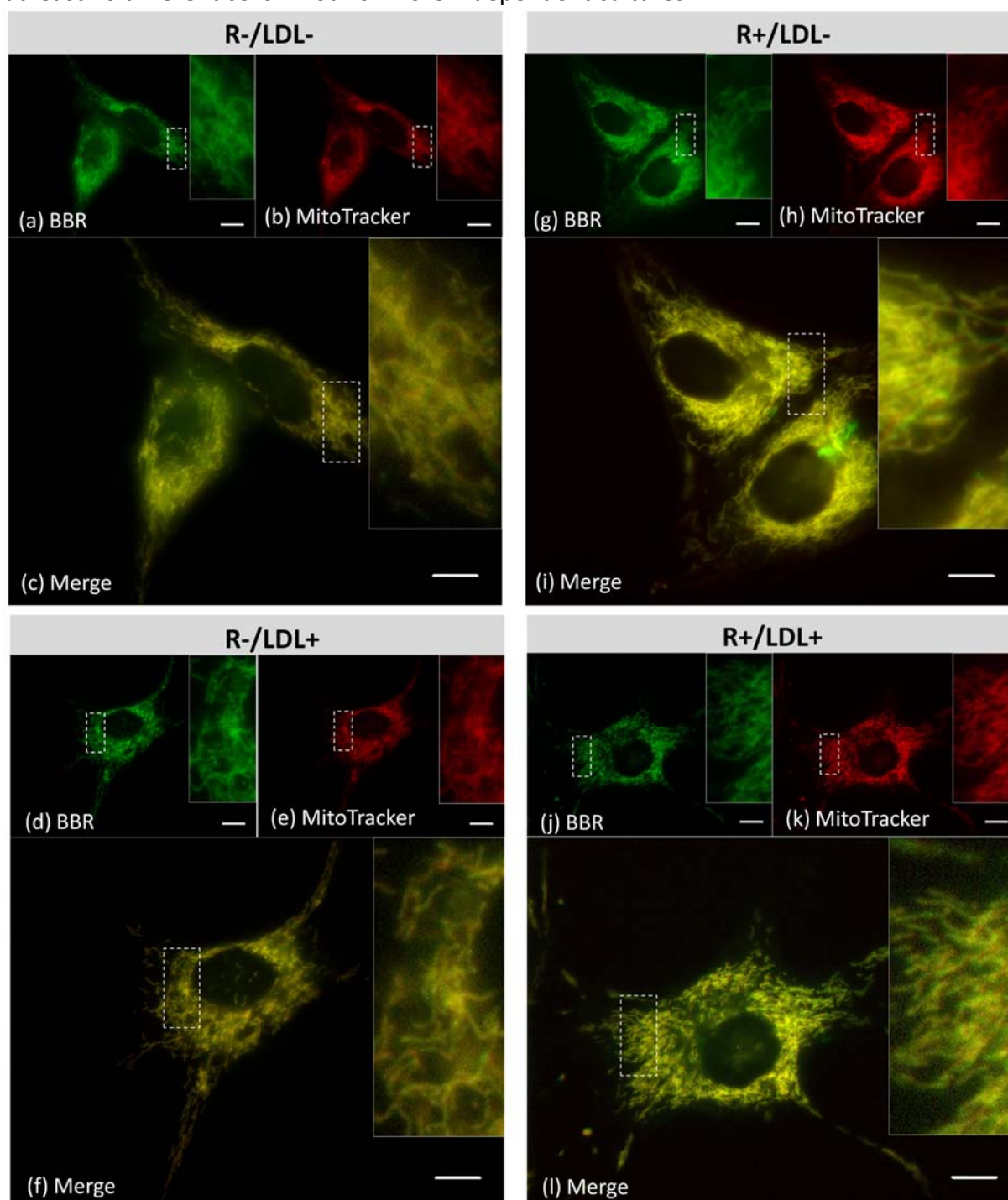
**Figure 3: Quantification of the berberine binding to LDL followed by changes of LDL fluorescence - Class P fixation.** (excitation 280 nm) (A) Quenching of LDL fluorescence ( $[LDL] = 2 \times 10^{-8} M$ ) upon addition of increasing amounts of berberine.  $[berberine]/[LDL] = 0$  to 1000 in the arrow direction. (B) Stern-Volmer plot. (C) Relative fluorescence intensity at 330 nm as a function of  $[berberine]/[LDL]$ .  $[LDL] = 2 \times 10^{-8} M$  (black circles) or  $6 \times 10^{-8} M$  (white circles). (D) Scatchard plot obtained from computed data (Halfman and Nishida, 1972).



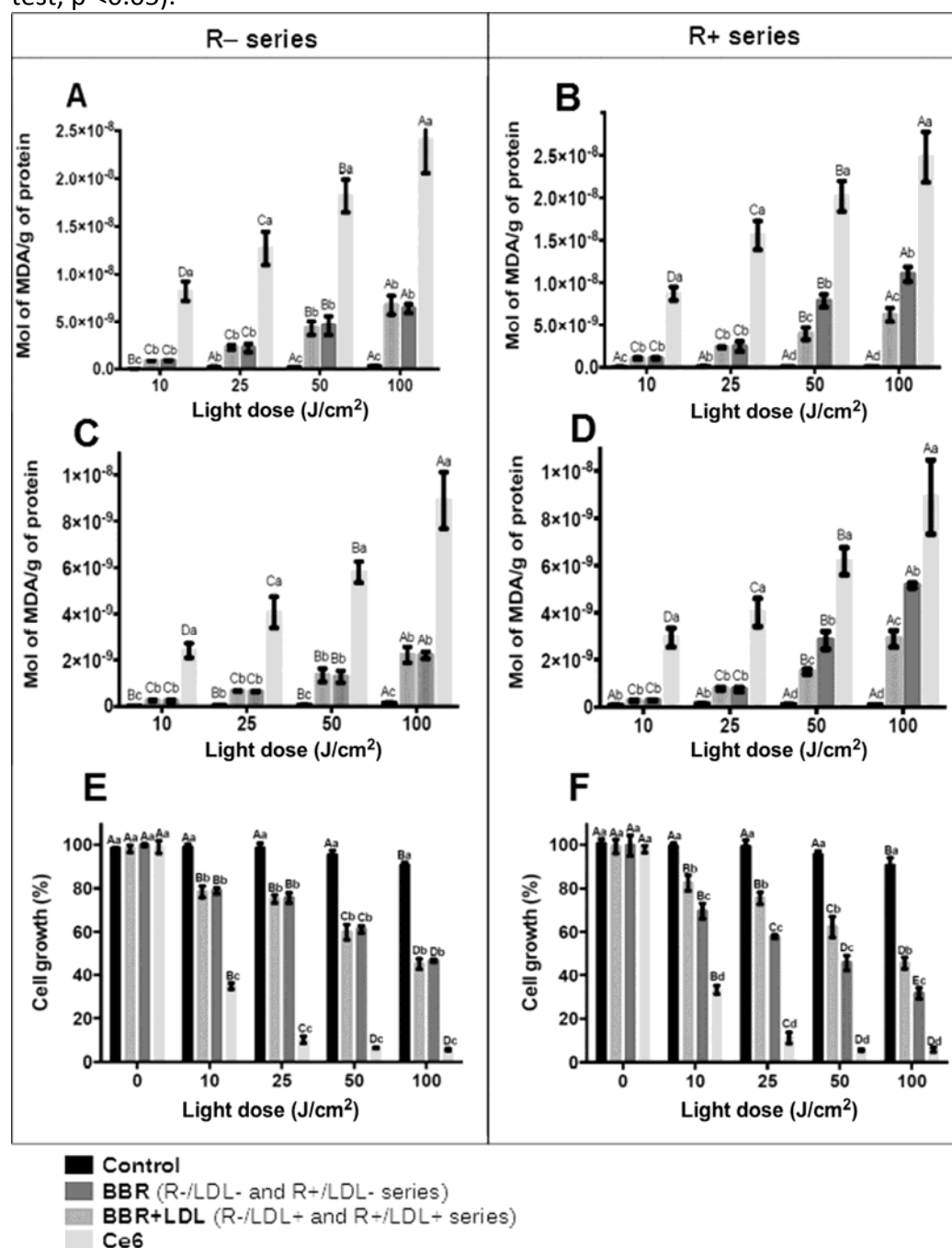
**Figure 4: Berberine uptake by human primary glioma cells.** Cells were treated according with series R-/LDL-, R-/LDL+, R+/LDL- and R+/LDL+: Berberine was preloaded or not in LDL and incubated with cells over-expressing or not the LDL-receptors. (*Top*) Effects of LDL receptor expression on the uptake of berberine. Statistical differences are indicated at the top of histogram bars by letters. Significant differences of berberine amounts between treatments within the same incubation time are indicate by lowercases a,b,c. The significant ones obtained between different incubation time with the same treatment are indicate using uppercases A,B,C. (one-way ANOVA followed by Tukey test,  $p < 0.05$ ). (*Bottom*) Fluorescence microscopy of U87MG cells. The intracellular fluorescence of berberine was collected as described in the material and method section 2.5 (Excitation  $355 \pm 25$  nm, emission  $535 \pm 25$  nm). Incubation time: 60 minutes. R-/LDL- (a), R-/LDL+ (b), R+/LDL- (d), R+/LDL+ (e). The third column (c, f) shows intrinsic fluorescence emission of U87MG control cells. Scale bars 5  $\mu$ m.



**Figure 5: Fluorescence microscopy study of the incorporation of berberine into U87MG glioma cells.** Cells were treated with berberine according with series R-/LDL- (a, b, c), R-/LDL+ (d, e, f), R+/LDL- (g, h, i) and R+/LDL+ (j, k, l) and stained with MitoTracker Red. For each treatment, the green fluorescence of the berberine (a, d, g, j) and the red fluorescence of the MitoTracker red (b, e, h, k) are shown as well as the merge image of these two color (c, f, i, l). Yellow codes for colocalization. Inset are enlarged details of the figures. Scale bars 10  $\mu$ m. In this and all subsequent experiments, typical traces or images are representative of at least 20 different cells in four or more independent cultures.



**Figure 6: Photo-induced effect of berberine on glioma cell line.** (Top and Middle). Lipid peroxidation within R- cells (A, C) and R+ cells (B, D). MDA (moles per gram of protein) was measured in cell extract (A, B) and supernatant (C, D). Lowercases and uppercases indicate statistical differences of MDA dose between treatments (LDL+ or LDL-) within the same light dose and between the same treatment in different light dose, respectively (one-way ANOVA followed by Tukey test,  $p < 0.05$ ). No significant statistical difference were detected between R-/LDL- and R-/LDL+ within the same light-dose. (Bottom). Cell viability (MTT assay) of U87MG cells. R- series (E) and R+ series (F). Lowercases and uppercases indicate statistical differences of cell viability between treatments within the same light-dose and between the same treatment in different light-dose, respectively (one-way ANOVA followed by Tukey test,  $p < 0.05$ ).



## Tables

**Table 1. Values of the equilibrium constants of the association of the berberine with the low density lipoproteins.** <sup>(1)</sup> Values calculated from the data obtained by following the changes in intrinsic LDL fluorescence. <sup>(2)</sup> Values calculated from data obtained by following the berberine fluorescence.

<b>K<sub>LDL</sub></b>					
1.76×10 <sup>8</sup> M <sup>-1(2)</sup>					
<b>K<sub>P</sub> ( = K<sub>P1</sub> + K<sub>P2</sub>)</b>				<b>K<sub>L</sub></b>	
1.65×10 <sup>8</sup> M <sup>-1</sup> (1.55×10 <sup>8</sup> M <sup>-1</sup> + 1.00×10 <sup>7</sup> M <sup>-1</sup> ) <sup>(1)</sup>				1.10×10 <sup>7</sup> M <sup>-1 (1,2)</sup>	
<b>n<sub>P1</sub></b>	<b>K<sub>P1i</sub></b>	<b>n<sub>P2</sub></b>	<b>K<sub>P2i</sub></b>	<b>n<sub>L</sub></b>	<b>K<sub>Li</sub></b>
180 <sup>(1)</sup>	8.6×10 <sup>5</sup> M <sup>-1(1)</sup>	40 <sup>(1)</sup>	2.5×10 <sup>5</sup> M <sup>-1(1)</sup>	~180 <sup>(1,2)</sup>	~6×10 <sup>4</sup> M <sup>-1 (1,2)</sup>